SAFROLE AS STARTING MATERIAL FOR THE SYNTHESIS OF 3-ALKYL MUCONIC ACID MONOMETHYL ESTERS AND 4,4-DIALKYL BUTENOLIDES

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ABSTRACT - Compounds <u>1b-d</u>, prepared from safrole (<u>1a</u>), were submitted to ozonation in methanol at 0° C furnishing regioespecifically the 3-alkyl muconic acid monomethyl esters (<u>2b-d</u>). These were transformed into the corresponding 4,4-dialkylated butenolides (3b-d).

As part of a program of study of abundant natural products available in Brazil we turned our attention to the transformation of safrole (la) into aliphatic derivatives. Thus compounds 1b-d were prepared from la by usual methods and were submitted to ozonation in methanol at 0° C. Under these conditions acidic products were obtained. After base extraction these products were characterized as the 3-alkyl muconic acid monomethyl esters (2b-d)2. ozonation of veratrolic systems is known in the literature3, this paper is, to our knowledge, the first to report about ozonation of aromatic compounds substituted by the dioxolane group4. The total control of the stereochemistry of the double bonds and the observed regiospecificity with respect to the position of the ester and functions makes these reactions very attractive. 2 structures of 2b-d were confirmed by their cyclization, catalyzed by silica gel5, to the corresponding alkyl butenolides (3b-d).

Since the butenolide moiety is present in many natural products and some of them exhibit interesting biological activities, there is a continuing need for simple and versatible synthetic methods for this class of compounds⁶. With the present approach possible to prepare regiospecifically 4,4-dialkylated butenolides startting from safrole derivatives in 63% overall yield.

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- \underline{b} ; R= CH₂CH₂CH₃
- c; R= CH2CH2OAc
- d; R= CH2CH2CH2OAc

References and notes

- P.R.R. Costa, C.C. Lopes and A.V. Pinto, Synth. Comm., 13, 691 (1983); P.R.R. Costa and A.J. Rabi, J. Chem. Research(S), 22 (1984); P.R.R. Costa, J.A. Rabi and W.M. Queiroz, J. Chem. Research(S), 20 (1984); E.J. Barreiro, P.R.R. Costa, F.A.S. Coelho and F.M.C. Farias, J. Chem. Research(S) 220 (1985); J. Chem. Research(M), 2301 (1985).
- The oxidation of cathecols with Cu^I salts gives mixtures of muconic acid monomethyl esters. See reference 5.
- R.B. Woodward, M.P. Cava, W.D. Ollis, A. Hunger, H. U. Daeniker and K. Schenker, <u>Tetrahedron</u>, <u>19</u>, 247 (1963).
- 4. For the oxidation of veratrolic and methylenedyoxi derivatives with ozone see P.R.R. Costa, S. Pinheiro and C.C. Lopes, Tetrahedron Lett., 4155 (1985).
- 5. T.R. Demmin, M.D. Swerdloff and M.M. Rogić, J. Am.

 Chem. Soc., 103, 5795 (1981); M.M. Rogić and T.R.

 Demmin, J. Am. Chem. Soc., 100, 5472 (1978).
- R. Okazaki, Y. Negishi and N. Inamoto, <u>J. Org. Chem.</u>,
 49, 3819 (1984).